

Dictyol A and B, Two Novel Diterpene Alcohols from the Brown Alga *Dictyota dichotoma*

By ERNESTO FATTORUSSO,* SILVANA MAGNO, LUCIANO MAYOL, CIRO SANTACROCE, and DONATO SICA
(*Istituto di Chimica Organica, Università di Napoli, Napoli, Italy*)

and VINCENZO AMICO, GIOVANNA ORIENTE, MARIO PIATTELLI, and CORRADO TRINGALI
(*Istituto di Chimica Organica, Università di Catania, Catania, Italy*)

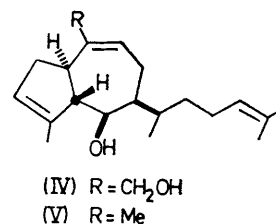
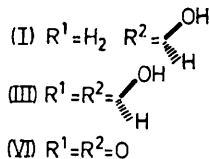
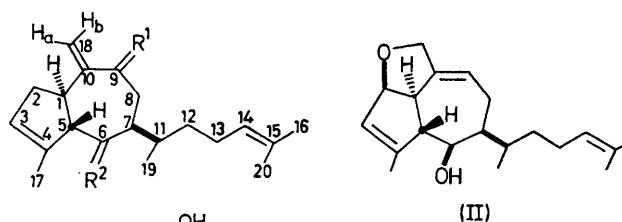
Summary From the brown alga *Dictyota dichotoma*, two diterpene alcohols with a hydroazulene skeleton, dictyol A (II) and B (III), have been isolated and their structures determined on the basis of spectroscopic and chemical evidence.

In 1973 Hirschfeld *et al.*¹ reported the isolation of the diterpene alcohol pachydictyol A (I) from the brown marine alga *Pachydictyon coriaceum* (Dictyotaceae). This compound, which possessed a mild antibiotic activity, was unique among diterpenes in having a hydroazulene skeleton.

During a survey of Mediterranean seaweeds,² we observed that the allied species *Dictyota dichotoma* var. *implexa* (Desf.) J. Ag. contains two metabolites whose spectral properties are strongly reminiscent of those of pachydictyol A. These compounds, dictyol A (II) and B (III) were isolated (in *ca.* 0.042 and 0.034% yield of the fresh plant, respectively) from the chloroform extract of the alga by conventional column chromatography (silica gel; CHCl₃-MeOH 99:1).

Dictyol A (II), [α]_D + 86.2°, m.p. 85–86 °C, has the molecular formula C₂₀H₃₀O₂ (*M*⁺, *m/e* 302.2249, calc. 302.2246). The i.r. spectrum (ν_{\max} 3400 cm⁻¹) showed the presence of an OH group. The n.m.r. spectrum (90 MHz, CDCl₃) displayed signals at δ 0.99 (3H, d, *J* 6 Hz, 11-Me), 1.61 (3H, s, *trans*-15-Me), 1.68 (3H, s, *cis*-15-Me), 1.86 (3H, s, 4-Me), 2.85br (1H, d, 5-H), 3.13 br (1H, 1-H), 4.06 (1H, m, 6-H), 4.36br (2H, s, 18-H₂), 4.62br (1H, d, *J* 7 Hz, 2-H), 5.09br (1H, t, *J* 7 Hz, 14-H), and 5.46br (2H, 3- and 9-H). The mass spectrum showed prominent peaks at *m/e* 284, 273, 269, and 255. Reduction of (II) with Na-NH₃ in EtOH at -45 °C gave (IV) as the main product, [α]_D - 17.0°; m.p. 92–94 °C, ν_{\max} 3320 cm⁻¹ (OH), δ

(CDCl₃) 0.97 (3H, d, *J* 6 Hz, 11-Me), 1.59 (3H, s, *trans*-15-Me), 1.67 (3H, s, *cis*-15-Me), 1.83 (3H, s, 4-Me), 3.81–4.20br (3H, 6-H and 18-H₂), 5.10br (1H, t, *J* 7.5 Hz, 14-H), 5.33 (1H, m, 3-H), and 5.66 (1H, m, 9-H); *m/e* 304, 286, 273, 271, 268, 255, and 253. In addition, small amounts of (V) {oil, [α]_D - 17.60°, *M*⁺, *m/e* 288; ν_{\max} 3500 cm⁻¹ (OH), δ (CDCl₃) 0.96 (3H, d, *J* 6 Hz, 11-Me), 1.59 (3H, s, *trans*-15-Me), 1.68 (6H, s, 10-Me and *cis*-15-Me), 1.83 (3H, s, 4-Me), 3.87 (1H, m, 6-H), 5.06br (1H, t, 7 Hz, 14-H), 5.23–5.56br (2H, 3- and 9-H)} and pachydictyol A



were also isolated. All the above data are consistent with structure (II) for dictyol A. The stereochemistry at C-2 was deduced from the fact that closure of the tetrahydrofuran ring requires a *cis* relationship between 1- and 2-H.

Dictyol B (III), $[\alpha]_D + 73.5^\circ$, m.p. 110–114 °C, has molecular formula $C_{20}H_{32}O_2$ (M^+ , m/e 304.2400; $C_{20}H_{32}O_2$ requires 304.2402), i.r. absorptions at ν_{max} 3340 (OH), 3040, 1640, and 896 cm^{-1} ($>C=CH_2$); δ (CCl_4) 1.00 (3H, d, J 6 Hz, 11-Me), 1.59 (3H, s, *trans*-15-Me), 1.67 (3H, s, *cis*-15-Me), 1.75 (3H, s, 4-Me), 3.65–4.03br (2H, 6- and 9-H), 4.81br (1H, s, H_a), 5.05br (1H, t, J 7 Hz, 14-H), 5.08br (1H, s, H_b), and 5.21 (1H, m, 3-H); m/e 286 and 268. Reduction of dictyol B with Na-NH₃ in EtOH at -45 °C gave (V), while Jones' oxidation furnished the diketone

(VI), as an oil, $[\alpha]_D - 6.6^\circ$, M^+ , m/e 300; ν_{max} 1704 (seven-membered ring ketone) and 1692 cm^{-1} ($\alpha\beta$ -unsaturated ketone); δ (CCl_4) 1.02 (3H, d, J 6 Hz, 11-Me), 1.53 (3H, s, *trans*-15-Me), 1.61 (3H, s, *cis*-15-Me), 1.71 (3H, s, 4-Me), 4.90br (1H, t, J 7 Hz, 14-H), 5.19br (1H, s, H_a), 5.40br (1H, s, 3-H), and 6.01br (1H, s, H_b).

From these data, structure (III) was deduced for dictyol B; the stereochemistry at C-9 has been tentatively assigned on the basis of further examination of the n.m.r. spectrum in the presence of variable amounts of Eu(fod)₃.

This work was supported by the C.N.R. (Italy).

(Received, 15th April 1976; Com. 426.)

¹ D. R. Hirschfeld, W. Fenical, G. H. Y. Lin, R. M. Wing, P. Radlick, and J. J. Sims, *J. Amer. Chem. Soc.*, 1973, **95**, 4049.

² G. Impellizzeri, S. Mangiafico, G. Oriente, M. Piattelli, S. Sciuto, E. Fattorusso, S. Magno, C. Santacroce, and D. Sica, *Phytochemistry*, 1975, **14**, 1549; E. Fattorusso, S. Magno, C. Santacroce, D. Sica, G. Impellizzeri, S. Mangiafico, G. Oriente, M. Piattelli, and S. Sciuto, *ibid.*, p. 1579; E. Fattorusso, S. Magno, L. Mayol, C. Santacroce, D. Sica, V. Amico, G. Oriente, M. Piattelli, and C. Tringali, *Tetrahedron Letters*, 1976, 937; E. Fattorusso, S. Magno, C. Santacroce, D. Sica, B. Di Blasio, C. Pedone, G. Impellizzeri, S. Mangiafico, G. Oriente, M. Piattelli, and S. Sciuto, *Gazzetta*, in the press.